



# Binding without folding – the biomolecular function of disordered polyelectrolyte complexes

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Recent evidence shows that oppositely charged intrinsically disordered proteins (IDPs) can form high-affinity complexes that involve neither the formation of secondary or tertiary structure nor site-specific interactions between individual residues. Similar electrostatically dominated interactions have also been identified for positively charged IDPs binding to nucleic acids. These highly disordered polyelectrolyte complexes constitute an extreme case within the spectrum of biomolecular interactions involving disorder. Such interactions are likely to be widespread, since sequence analysis predicts proteins with highly charged disordered regions to be surprisingly numerous. Here, we summarize the insights that have emerged from the highly disordered polyelectrolyte complexes identified so far, and we highlight recent developments and future challenges in (i) establishing models for the underlying highly dynamic structural ensembles, (ii) understanding the novel binding mechanisms associated with them, and (iii) identifying the functional consequences.

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## Introduction

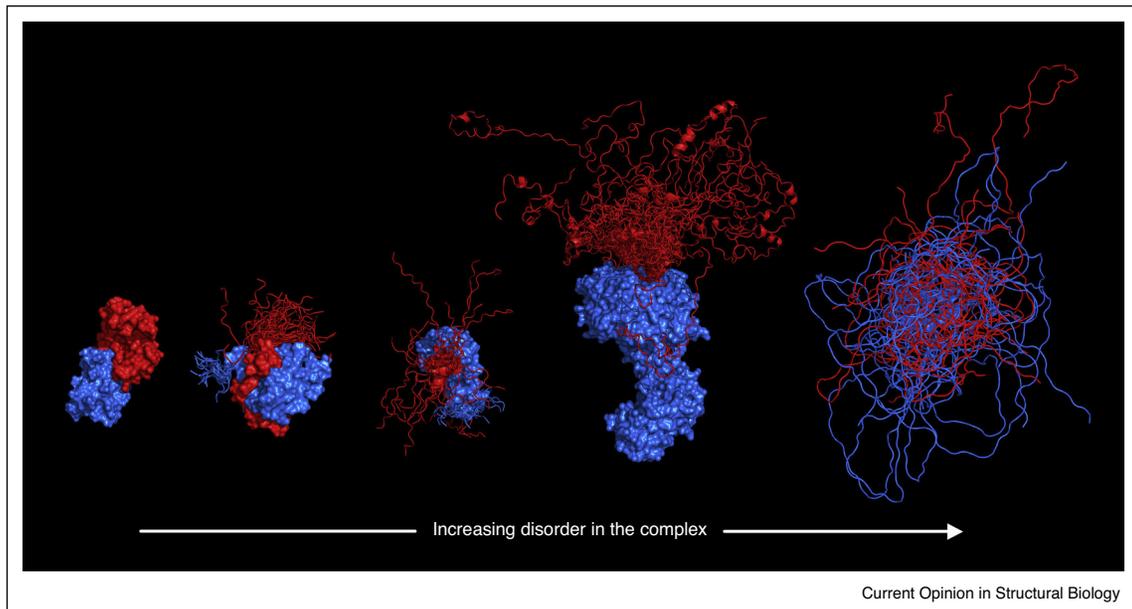
The textbook view of biomolecular interactions typically involves structured macromolecules with well-defined

interfaces whose precise complementarity enables the tight and specific binding required for biomolecular function. This notion was shaped by decades of progress in structural biology that has revealed atomically detailed views of thousands of complexes between proteins, nucleic acids, and other biomolecules. Considering this information, we now understand many of the mechanisms that drive biomolecular recognition, self-assembly, and cellular function. However, there has been evidence for at least three decades that this notion is incomplete. Already in his 1988 classic “Acid blobs and negative noodles”, Paul Sigler pointed out that studies of highly charged protein segments in transcriptional regulation “suggest a disquieting picture of a conformationally ill-defined polypeptide that can function almost irrespective of sequence” [1].

The following decades revealed a wide spectrum of biomolecular complexes involving intrinsically disordered proteins (IDPs), sometimes referred to as “fuzzy” complexes [2] (Figure 1). Among the earliest examples were transcription factors that fold upon binding to DNA [3]. This concept of coupled folding and binding was soon extended to protein–protein interactions [4]. The degree of folding can, however, vary widely: In some cases, the IDPs form extensive secondary and tertiary structure upon binding; in other cases, only short segments are involved that either form secondary structure (most commonly  $\alpha$ -helices) or bind in extended form (comparable to peptide binding by antibodies), but large parts of the sequence remain disordered [2]. It then became clear that if multiple such binding segments are present in the IDP, their rapid association and dissociation with specific interaction sites on the folded binding partner can result in highly dynamic complexes where the IDP stays almost completely disordered and forms multivalent, rapidly exchanging interactions involving only transient local ordering [5–11]. However, even in such cases, the site-specific interactions governing the association require van-der-Waals contact and generally form structurally defined complementary interfaces, albeit small and potentially short-lived [7]. But can proteins form functionally relevant complexes even in the complete absence of such structurally defined and site-specific interactions?

Answering this question has been difficult and controversial, at least in part because of the lack of suitable experimental

Figure 1



The order-to-disorder spectrum of protein complexes. The traditional view of protein binding has focused on the interactions between folded proteins based on highly complementary, well-structured interfaces. It has become increasingly clear, however, that proteins can retain disorder in their bound states [2]. The increasing levels of disorder are illustrated here with the following examples from left to right: Colicin E9 with cognate Im9 (crystal structure, PDB 1EMV); RelA-TAD/CBP-TAZ1 complex (NMR structure, PDB 2LWW); Gcn4 activation domain bound to the mediator co-activator domain 1 of Gal11/med15 (NMR structure, PDB 2LPB); complex of Sic1 with the Cdc4 subunit of ubiquitin ligase (ensemble based on NMR data [98]); complex of H1 and ProT $\alpha$  (ensemble from coarse-grained simulations based on single-molecule FRET data [18\*\*]). Highly disordered polyelectrolyte complexes constitute the extreme case, where both binding partners can fully retain their disorder. Stably folded proteins or regions are shown in surface representation. Disordered regions are shown in cartoon representation, with multiple conformations from the NMR structure or the modeled ensemble overlaid. (Figure prepared using PyMOL, Schrödinger).

methods and theoretical concepts. One of the earliest suggestions of a complex between IDPs involving no formation of structure, the weak homo-oligomerization of the cytoplasmic domain of the T-cell receptor  $\zeta$  chain [12], could not be confirmed independently [13]; but since then, more evidence for highly disordered complexes has accumulated. Thomas *et al.* [14] presented evidence for the interaction between linker histones and the high-mobility group protein B1 (HMGB1) mediated predominantly by the charged disordered regions of the proteins that involved no detectable formation of secondary structure. Recently, the molten-globule-like C-terminal domain of the adapter protein 4.1 has been reported to interact with a disordered 26-residue peptide from the nuclear mitotic apparatus protein without the formation of stable structure [15\*]. Individual amino acid exchanges identified based on molecular dynamics (MD) simulations, especially in hydrophobic stretches of both binding partners, were able to abolish the micromolar binding and thus suggest rather site-specific interactions, but apparently without the formation of a persistent interface [15\*]. Evidence for such disordered complexes has also emerged for the interaction of the disordered chromatin protein NUPR1 and its paralogue NUPR1L with several disordered binding partners, including the nuclear protein prothymosin  $\alpha$  and the histone

acetyl transferase-associated protein MSL1 [16,17]; again, several hydrophobic residues seem to play a key role for binding, indicating site-specificity, albeit in the absence of structure formation.

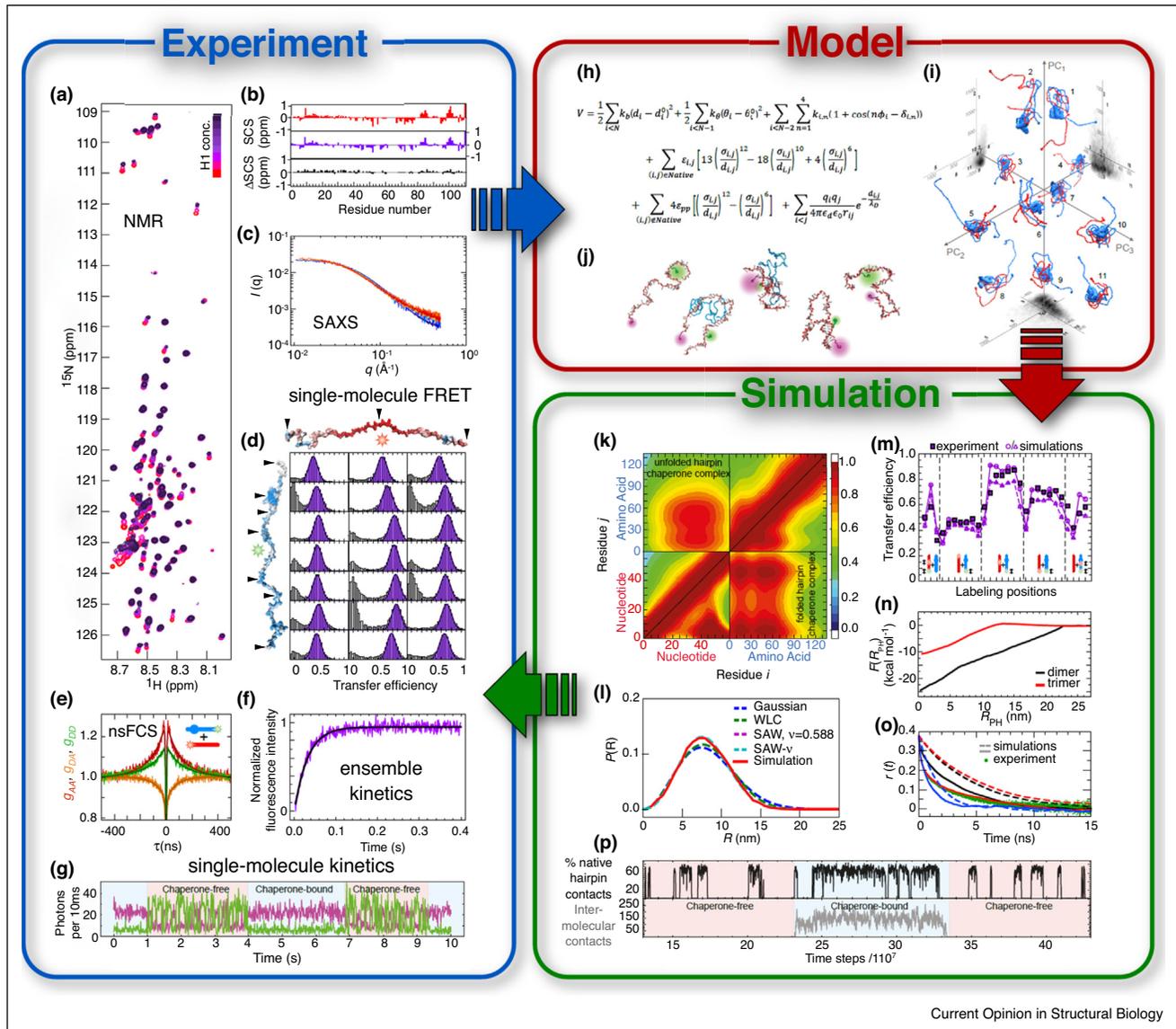
### Highly disordered polyelectrolyte complexes of IDPs

An extreme case of an IDP complex showing neither signs of site-specific interactions nor structure formation is that of the linker histone H1.0 (H1) and its nuclear chaperone prothymosin  $\alpha$  (ProT $\alpha$ ) [18\*\*]. H1 is an IDP involved in chromatin condensation by binding to nucleosomes [19,20]; it is highly positively charged and comprises two disordered regions flanking a small folded domain [21]. The abundant nuclear protein ProT $\alpha$  is a fully disordered and highly negatively charged IDP [22,23] involved in chromatin remodeling [24], transcription, cellular proliferation, and oncogenesis [25]. ProT $\alpha$  had previously been shown to bind to H1 *in vitro* [26] and *in vivo* [27] and to act as a linker histone chaperone that increases the mobility of H1 in the nucleus [27]. However, given the large fraction of charged residues and the concomitant scarcity of hydrophobic amino acids in both IDPs, the mode of interaction between these two biological polyelectrolytes was entirely unclear.

Circular dichroism and NMR spectroscopy revealed no detectable formation of secondary or tertiary structure upon binding (Figure 2a,b). The only clear signatures of binding from NMR were slight changes in chemical shifts, resonance peak intensities, and relaxation times,

which were most pronounced for residues in regions with the highest charge density. Nevertheless, single-molecule Förster resonance energy transfer (FRET) spectroscopy demonstrated a dissociation constant in the low picomolar range at a near-physiological ionic

Figure 2



Elucidating the conformational distributions, dynamics, and functions of highly disordered complexes with integrative modeling and simulations. The combination of complementary experimental techniques enables different aspects of structure and dynamics to be probed, for example residue-specific changes in conformation, secondary and tertiary structure from NMR (a,b), overall dimensions from small-angle X-ray scattering SAXS (c), segment-specific intramolecular and intermolecular distance distributions from single-molecule FRET (d), long-range dynamics from nsFCS (e), kinetics of binding and conformational change from ensemble (f) and single-molecule experiments (g). Experimental data can inform the choice of a suitable physical description, such as analytical models from polymer physics and coarse-grained or atomistic simulations. A versatile approach illustrated here are coarse-grained simulations, which combine residue-level interactions (h) with a broad range of accessible timescales, enabling, for example the investigation of structural ensembles (i, principal component analysis of the H1-ProT $\alpha$  complex) and interaction mechanisms (j), folding of single-stranded nucleic acids in the presence of NCD). Simulations can yield a wide range of information, including contact maps (k), intramolecular and intermolecular distance distributions (l), transfer efficiencies (m), interaction potentials of mean force (n), mobility of fluorophores (o), and binding kinetics (p). By quantitative comparison with experiments and parameter optimization, the model can be improved iteratively to achieve a realistic description of the system. Panels a, b, d, e, f, h, i, m, n are from Borgia *et al.* [18 $^{**}$ ], panels g, j, k, p from Holmstrom *et al.* [46 $^{**}$ ], panel c from Borgia *et al.* [99], panel l from Zheng *et al.* [100], and panel o from Best *et al.* [101].

strength of 165 mM. A mere doubling of the ionic strength, however, reduced the affinity to the micromolar range, and experiments with fragments of H1 showed that the highly charged disordered C-terminal region made by far the largest contribution to binding, reflecting the dominance of electrostatic interactions [18\*\*]. In contrast, the small folded domain of H1 was dispensable for high affinity. Measurements with different fluorophores and labeling positions [18\*\*] and competition experiments with unlabeled protein (Sottini *et al.*, unpublished) demonstrate that the fluorophores have a negligible effect on the affinity. Nanosecond fluorescence correlation spectroscopy (nsFCS, Figure 2e) [28] revealed pronounced long-range distance dynamics in the complex on the 100-ns timescale, comparable to the chain reconfiguration times of the free proteins and other IDPs [28,29]. All experimental data thus pointed towards a highly disordered and dynamic protein complex. A key step towards understanding the underlying interaction mechanism was to describe the structurally diverse ensemble with a physics-based model.

The integrative modeling of conformational ensembles based on the combination of experimental data with theory and simulations (Figure 2) has made remarkable progress and ranges from analytical theory and simple models to all-atom MD simulations [30\*,31\*,32–37]. Given the large size of the H1-ProT $\alpha$  complex, a coarse-grained model turned out to provide the best compromise between molecular detail and computational feasibility [18\*\*,38,39]. Each amino acid residue was represented as a single bead with appropriate volume and charge; electrostatic interactions were described by a Coulomb potential including Debye-Hückel screening; and the folded domain of H1 was stabilized by a structure-based potential. A particularly appealing aspect of this approach is that all parameters of these force field terms are uniquely defined by the model. The single adjustable parameter remaining was the strength of the Lennard-Jones term, which models all remaining short-range interactions and was chosen to be the same for all residues; it could be quantified by globally optimizing the agreement of the simulations with 28 intramolecular and intermolecular single-molecule FRET efficiencies reporting on distributions of distances between pairs of residues (Figure 2d,m) [18\*\*]. The resulting simulations illustrate the extreme disorder in the complex (Figures 1 and 2i) and reproduced the high affinity, the diffusion-limited binding kinetics, and the NMR data: The smooth and broad distribution of the average number of contacts made by the residues of ProT $\alpha$  closely resembled the distribution of changes in chemical shifts, peak intensities, and relaxation times upon binding. This comparison provided an independent benchmark of the model and suggests that nonspecific, electrostatically dominated interactions can account for the high affinity of the disordered H1-ProT $\alpha$  complex, reminiscent of a mean-field picture [5], without the need for the persistent

site-specific interactions required in the classical view of multivalency and related models.

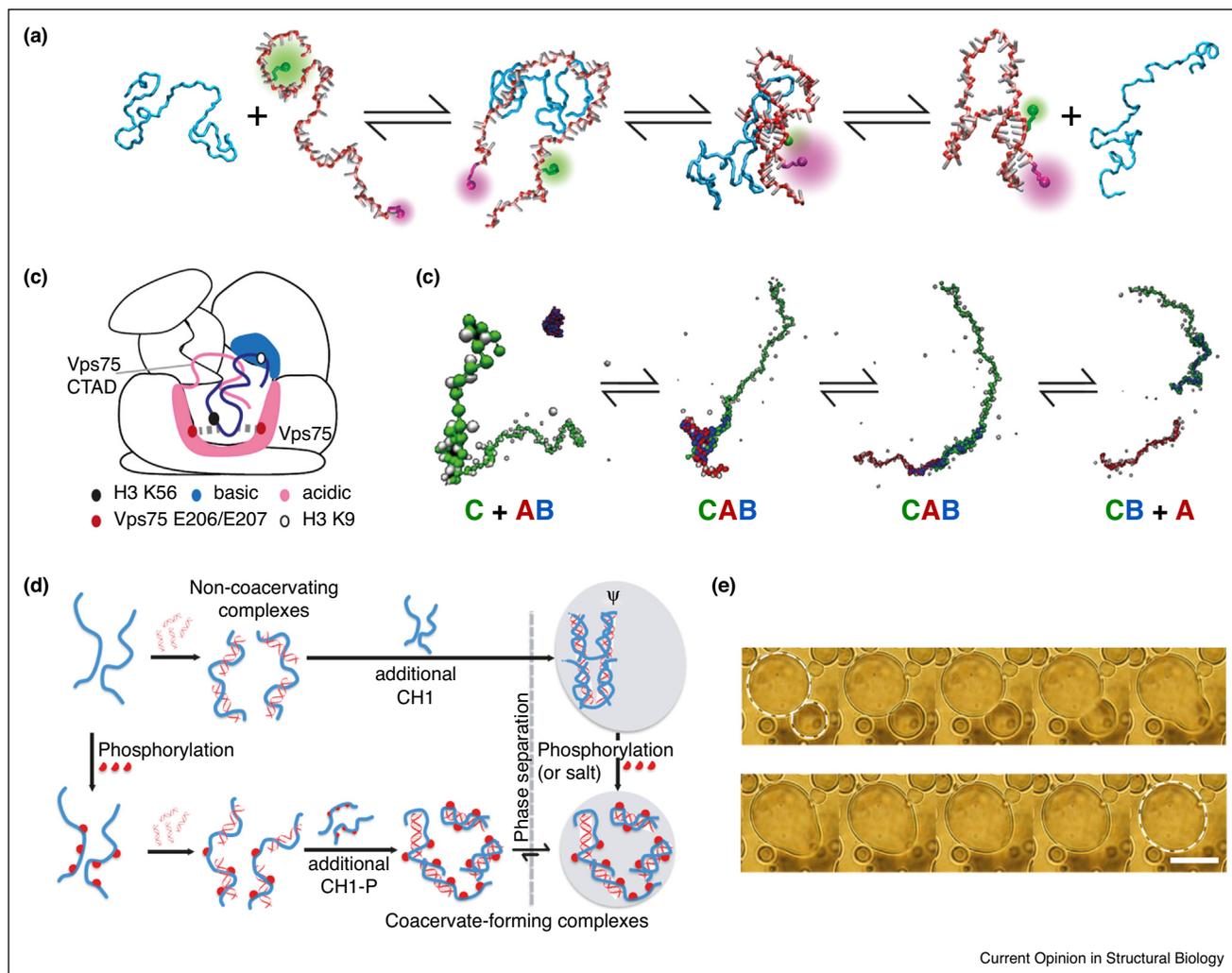
Recently, another example of this new paradigm of disordered polyelectrolyte interactions with interesting functional implications has been demonstrated for the histone chaperones Asf1 and Vps75 with the Core histone dimer H3:H4 and Rtt109, an enzyme responsible for lysine acetylation in H3 [40\*\*]. Employing a combination of NMR, small angle neutron scattering, MD simulations, and integrative modeling, the authors found that one of the disordered and negatively charged C-terminal tails of Vps75 establishes electrostatic interactions with the disordered N-terminal tail of H3 in the cavity of the complex, where the catalytic site of Rtt109 is located (Figure 3b). Remarkably, the electrostatic association does not induce the formation of detectable structure. The authors propose that this type of interaction helps to localize the lysine residues in the H3 tail close to the catalytic site of Rtt109 with minimal loss of conformational entropy and maintains full accessibility for acetylation, a compelling mechanism that may be widespread in the posttranslational modification of IDPs.

### Highly disordered polyelectrolyte complexes between IDPs and nucleic acids

Considering the nature of highly disordered complexes formed by charged IDPs, it seems plausible that similar kinds of polyelectrolyte complexes can be formed between positively charged IDPs and nucleic acids. Indeed, two cases of this type have recently been reported. The first one involves the disordered C-terminal tail of histone H1.11L (CH1), which exhibits nanomolar affinity to double-stranded DNA comparable in length to the DNA linkers between nucleosomes [41\*\*]. As demonstrated by NMR spectroscopy, CH1 is not only disordered in the free state but also when bound to DNA (Figure 3d). As in the case of H1 binding to ProT $\alpha$ , the relatively small chemical shift perturbations upon binding were broadly distributed along the sequence, and neither secondary structure propensity core nor heteronuclear NOEs exhibited pronounced changes, suggesting that CH1 remains highly dynamic in the bound state and does not exhibit site-specific interactions with the DNA. A very important implication of this result is that the disordered tails of linker histones such as H1 are likely to remain disordered when bound to the nucleosome, an issue that has been challenging to address quantitatively [42,43]. Stott *et al.* further investigated the role of phosphorylation at three serine residues of CH1, which did not alter the interaction mechanism but led to a reduction in affinity, an aspect that is expected to be important for cellular regulation by posttranslational modifications [41\*\*].

Another recent example of a polyelectrolyte complex between an IDP and nucleic acids is the hepatitis C virus core protein, an intrinsically disordered RNA chaperone

Figure 3



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Highly disordered polyelectrolyte complexes expand the functional repertoire of biomolecular interactions. **(a)** The positively charged hepatitis C virus Core protein (blue) binds to nucleic acids (red) and locally screens the repulsive charge interactions with an efficiency equivalent to molar salt concentrations [46<sup>\*\*</sup>]. The resulting structure formation (hairpin folding in this example probed by single-molecule FRET) reflects the protein's function as an RNA chaperone. Figure adapted from Holmstrom *et al.* [46<sup>\*\*</sup>]. **(b)** The positively charged disordered tail of histone H3 (blue) forms polyelectrolyte interactions with the negatively charged disordered C-terminal region of its chaperone Vps75 (red), which confines the H3 tail in the central cavity of the complex and promotes the acetylation of specific lysine residues by increasing their proximity to the active site of the acetyltransferase Rtt109 [40<sup>\*\*</sup>]. Figure from Danilenko *et al.* [40<sup>\*\*</sup>]. **(c)** Coarse-grained Langevin dynamics simulations show how a polyanion chain invades the complex between two oppositely charged polyelectrolytes and replaces the polyanion by competitive substitution [80<sup>\*\*</sup>]. Figure adapted from Peng and Muthukumar [80<sup>\*\*</sup>]. **(d)** The polyelectrolyte interactions between the linker histone tail CH1 and DNA are modulated by ionic strength, protein and DNA concentrations, and phosphorylation, which can shift the balance between stoichiometrically defined complexes and coacervation [41<sup>\*\*</sup>]. Similar mechanisms may control nucleosome interactions and regulate chromatin compaction [41<sup>\*\*</sup>,42]. Figure from Turner *et al.* [41<sup>\*\*</sup>]. **(e)** The formation of coacervates from positively and negatively charged polyelectrolyte IDPs is a special case of the phase separation facilitated by IDPs [81,84]. The example shows condensed-phase droplet fusion at high concentrations of H1 and ProT $\alpha$  observed by video microscopy (scale bar 50  $\mu$ m; Borgia *et al.*, unpublished).

with positive net charge that promotes viral genome dimerization [44,45<sup>\*\*</sup>,46<sup>\*\*</sup>] (Figure 3a). Its function primarily involves interactions with single-stranded RNA to facilitate the formation of higher-order structure or larger assemblies [47]. Holmstrom *et al.* [46<sup>\*\*</sup>] investigated the interaction of the nucleocapsid domain (NCD) of the hepatitis C virus Core protein with several

single-stranded RNA and DNA sequences. Since NCD assembles into nucleocapsid-like particles with nucleic acids at the sample concentrations required for many ensemble techniques, such as circular dichroism or NMR spectroscopy [45<sup>\*\*</sup>,48], the 1:1 complex was probed at the very low concentrations accessible with single-molecule spectroscopy. nsFCS showed that the

long-range chain dynamics of free NCD on the 50-ns timescale are retained upon nucleic acid binding, indicating that it remains disordered. In simulations, performed using a coarse-grained model of the IDP similar to the one for H1-ProT $\alpha$  [18\*\*] and a three-bead-per-nucleotide model of the nucleic acid, both NCD and the single-stranded nucleic acid remained disordered in the complex (Figure 2j). The results showed very good agreement with intramolecular and intermolecular transfer efficiencies obtained from single-molecule FRET experiments and thus strongly support the notion that IDPs and nucleic acids can form highly disordered polyelectrolyte complexes with high affinity. In the case of NCD, the nanomolar binding to the highly flexible single-stranded RNA or DNA leads to a pronounced compaction of the nucleic acids via charge screening and can thus promote structure formation [46\*\*] (Figure 3a) – indeed the simulations showed that the acceleration of folding rate was entirely explained by the effect of collapsing the chain.

In summary, the discovery of such extremely disordered yet biologically functional complexes completes the missing upper boundary in the spectrum of biomolecular complexes that involve structural disorder (Figure 1) and further corroborates the importance of IDPs in biology. However, these systems also pose a wide range of new questions and scientific challenges, which are likely to reveal exciting new mechanisms in biomolecular assembly, regulation, and communication in the cell.

### Challenge I: structurally and dynamically realistic models

Moving forward, the first important challenge will be to advance the methods and concepts for developing realistic models of such highly disordered complexes. The description of IDPs in terms of highly dynamic and structurally diverse conformational ensembles has made tremendous progress in recent years, with models ranging from analytical polymer theory and coarse-grained representations to all-atom MD simulations [30\*,31\*,32,38,49\*,50]. Increasingly, integrative approaches are utilized that combine or benchmark a molecular model with experimental results, ideally from multiple complementary techniques. Many of these advances can be transferred directly to highly disordered complexes (Figure 2), but they will also require substantial further developments.

Analytical polymer models have been used very successfully for describing and understanding the physical properties of unfolded and intrinsically disordered proteins, including their intramolecular distance distributions and the response of chain dimensions to solution conditions [29,32,51]. A major advantage of such models is that they can be used for identifying key parameters responsible for the observed behavior by directly fitting experimental results. The theory of polyelectrolyte complexes, however, remains a

formidable challenge owing to the sequence-dependent long-range correlations within and between the chains and the coupling between polymer charge and counterions [52]. However, some earlier and recent approaches may lend themselves to suitable modifications for describing the conformational distributions within polyelectrolyte complexes [53–55,56\*].

At the other extreme in terms of detail are all-atom MD simulations. Impressive recent advances in force field development have started to enable faithful simulations of IDPs [30\*,31\*,32,57–63], which should also be applicable to polyelectrolyte complexes. However, given the lack of pronounced minima in the free-energy surfaces of IDPs and the correspondingly subtle balance of forces, fully transferable potentials have been difficult to establish, and adjustments of force field parameters may still be required, for example guided by experimental data [30\*,31\*]. Moreover, the computational cost for reaching the required lengthscales and timescales is often prohibitive, especially if solvent is included explicitly. The development of atomistic force fields for IDPs combined with implicit solvent is therefore of great interest [30\*,32,64]. Since such simulations have been shown to provide important insights into the single-chain properties of highly charged IDPs [65–67], they are also promising for describing polyelectrolyte complexes.

A third approach of intermediate complexity is the use of coarse-grained models [68] (Figure 2h–j). Even without relying on atomic resolution, they can capture essential aspects of the system, such as the heteropolymeric nature of biomolecules and the corresponding residue-characteristic non-covalent interactions within and between chains, including the electrostatics that dominate polyelectrolyte complexes [18\*\*,46\*\*,49\*,69–73]. A major advantage of these models is their lower computational cost, which enables model optimization and conceptual insight by systematic parameter variation and comparison to experimental observables that do not require atomic resolution, such as scattering data, FRET efficiencies, or patterns of NMR signal changes [18\*\*,38,46\*\*,68] (Figure 2). Moreover, coarse-grained simulations lend themselves to large systems, including complex biomolecular assemblies and even processes approaching mesoscopic scales, such as liquid-liquid phase separation [49\*,74\*]. Disadvantages of coarse-grained models are that some of the chemical differences between amino acids or nucleotides are difficult to capture, and that absolute timescales are not easily reproduced, although approximate rescaling can be achieved by comparison to experimental observables [75,76]. Important future developments may be an explicit representation of counterions and solvent, the optimization of residue-based potentials [77] to account for the heteropolymeric nature of proteins, and the extension to large biomolecular assemblies. Nevertheless, all-atom simulations are likely to remain indispensable for

addressing aspects that are sensitive to chemical structure, such as site-specific interactions, transient secondary structure formation, local relaxation phenomena that can be compared to NMR experiments [50], or the absolute timescales of dynamics [78].

### Challenge II: novel interaction mechanisms

The second major challenge will be to identify the mechanistic implications of highly disordered polyelectrolyte complexes and establish appropriate kinetic and thermodynamic binding models. For instance, even in cases where the complementarity of the charges and sizes of two binding partners leads to the preferential formation of 1:1 complexes, as in the case of H1-ProT $\alpha$ , the excess of one binding partner can result in larger oligomers [18\*\*]. This behavior is even more pronounced for systems with a substantial charge or size imbalance, as in the binding of CH1 [41\*\*] or NCD [46\*\*] to nucleic acids.

One important consequence of such higher-order complexes concerns the interaction kinetics. In the simplest and most commonly employed kinetic models for association-dissociation reactions, only binary complexes occur, and any complex needs to fully dissociate before another association event can take place. As a result, the average dwell time in a complex is independent of the concentration of free binding partner present in solution. Multivalency, however, (of which polyelectrolyte interactions could be considered a limiting case) can facilitate the formation of transient ternary complexes and concentration-dependent dwell times [79]. For H1-ProT $\alpha$ , such a mechanism has recently been shown to accelerate dissociation and exchange between binding partners by orders of magnitude (Sottini *et al.*, unpublished), which explains the previously observed transition from slow exchange between bound and unbound states at single-molecule conditions to fast exchange at NMR concentrations [18\*\*]. Coarse-grained simulations illustrate how the underlying process of competitive substitution [80\*\*] occurs by one polyelectrolyte chain replacing another via a transient ternary complex, and how it is facilitated by the highly dynamic nature of a disordered polyelectrolyte complex (Figure 3c). It is likely that competitive substitution has a pronounced influence on interaction kinetics in the cell, and together with the diffusion-limited association observed for polyelectrolyte complexes such as H1-ProT $\alpha$  [18\*\*], it may provide a mechanism of combining high-affinity interactions with rapid kinetics and cellular regulation on a biologically useful timescale.

A further important consequence of higher-order complex formation concerns the assembly of much larger structures, and ultimately macroscopic liquid-liquid phase separation. Research on biomolecular condensates over the past decade has shown that IDPs are often key

components for forming biomolecular condensates based on phase separation, such as the nucleolus or stress granules, frequently in combination with RNA [81]. Thus, it may not come as a surprise that polyelectrolyte systems such as CH1-DNA [41\*\*] (Figure 3d) or H1-ProT $\alpha$  (Figure 3e) exhibit liquid-liquid phase separation at sufficiently high concentrations. IDP polyelectrolyte complexes may thus provide an excellent opportunity for investigating the transition from the formation of oligomers with defined stoichiometry to phase separation. Notably, even for synthetic polyelectrolytes, whose association and the resulting phase separation, known as complex coacervation, have been investigated for almost a century [82,83], information about 1:1 complexes, small oligomers, and detailed structural and dynamic properties at the molecular level is largely lacking. Nonetheless, the concepts and methods from polymer and soft matter physics will be invaluable for investigating IDP complexes and phase separation [83–86].

In view of the promiscuity of charge interactions and the range of possible oligomerization states of polyelectrolyte complexes, a key question for future research will be whether, and if yes, how specificity of this type of biomolecular interactions can be achieved. This issue is particularly prominent in a cellular environment like the nucleus, which is crowded with highly charged proteins and nucleic acids. Polyelectrolyte interactions per se are unlikely to result in high specificity, although some selectivity for certain targets may be encoded in the chain length, the number or density of charges, and their distribution or patterning along the sequence, factors that also affect the conformational distributions within IDPs [32]. Another open question is the detailed interplay between Coulomb interactions and other contributions, for example from hydrophobic residues. Highly charged disordered regions often occur in combination with folded domains in nucleic-acid-binding proteins, where they can increase the affinity of the protein and facilitate its diffusion along the DNA or RNA [87], while specificity is encoded mainly in the folded domains. Similar combinations of folded and disordered binding modules with different roles in biomolecular recognition may also represent an effective and flexible way of modulating protein–protein interactions.

Furthermore, specificity can be affected by other biological regulation mechanisms, such as cellular localization or synchronized expression of binding partners during relevant stages of development or during the cell cycle. There might even be a role of local fluctuations in cellular ion concentrations [88], since ionic strength changes can have extreme effects on polyelectrolyte complex affinity [18\*\*,46\*\*,89]. It is noteworthy that the driving forces for polyelectrolyte complex formation are often dominated by the large entropy increase from the release of counterions and water, with only small enthalpic contributions,

either positive or negative [69,89,90]. Experimentally, the small reaction heats complicate investigations by isothermal titration calorimetry, and owing to additional signal contributions from oligomer formation or phase separation, a quantitative interpretation of the measurements can be challenging [41,91–93].

### Challenge III: functional implications

The third major challenge will be to clarify how widespread highly disordered polyelectrolyte complexes are in biology, which types of functions they perform, and in which way their functional repertoire extends beyond the classical concepts of biomolecular interactions. Since sequence analysis indicates that hundreds of proteins in the human proteome alone may participate in such interactions [18], it is likely that we have only seen the tip of the iceberg. Many of these proteins localize to the nucleus, and obvious candidates include the machinery involved in transcription, replication, and regulation, where positively charged disordered regions are highly abundant [42].

Interesting functional properties have started to emerge from the examples of polyelectrolyte complexes identified to date. Positively charged IDPs can act as macromolecular counterions that screen repulsive charges in nucleic acids with efficiencies equivalent to molar salt concentrations, as in the case of NCD [46], and in this way facilitate RNA or DNA structure formation (Figure 3a). Given the simple underlying physical principle, the same kind of mechanism is likely to affect many protein–nucleic acid interactions in the cell, including the positively charged tails of proteins involved in replication, transcription, and chromatin compaction [94]. On larger scales, the observation of phase separation, or complex coacervation, of highly charged IDPs and nucleic acids [41] (Figure 3d,e) indicates an important role for these mesoscopic assemblies, whose functional relevance is only starting to be understood [81,95]. Post-translational modifications, which are highly abundant in IDPs [96,97], are expected to regulate many of these processes, as illustrated by the sensitivity of phase separation of CH1 and DNA to phosphorylation [41] (Figure 3d). Intrinsic disorder is a way of facilitating access to modifying enzymes [96], and polyelectrolyte interactions can themselves contribute to these mechanisms, as illustrated by the intricate electrostatic interactions between the disordered N-terminal tail of histone H3 and the disordered negatively charged C-terminal tails of its chaperone Vps75 that promote lysine acetylation [40] (Figure 3b).

These exciting recent developments suggest that highly disordered polyelectrolyte complexes may be involved in many cellular processes and can further extend the growing functional repertoire of IDPs. The years ahead will provide ample opportunity for discovering novel molecular mechanisms that contribute to biological

function and go beyond the textbook notions of biomolecular interactions.

### Author contributions

All authors contributed to writing this paper.

### Conflict of interest statement

Nothing declared

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